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Key indicators

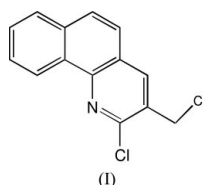
Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.060
 wR factor = 0.209
Data-to-parameter ratio = 21.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2-Chloro-3-chloromethylbenzo[*h*]quinolineThe title compound, $\text{C}_{14}\text{H}_9\text{Cl}_2\text{N}$, a derivative of benzo-
quinoline, is a tricyclic planar nitrogen heterocyclic system.

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Comment

The molecular structure of the title compound, (I), is composed
of a tricyclic planar nitrogen heterocyclic system, carrying two
substituents (Cl and CH_2Cl ; Fig. 1). This system displays
potential biological interest (Kerry *et al.*, 1999). The crystal
structure consists of planes stacked along the b axis. The
plane-to-plane separations are 3.46 and 3.50 Å (Fig. 2).

Experimental

The title compound was prepared according to the method of Meth-
Cohn *et al.* (1981). At a temperature ranging between 273 and 278 K,
0.07 mol of POCl_3 was added dropwise to 0.015 mol of dry N,N -
dimethylformamide. Stirring was continued for 30 min, then 0.01 mol
of 3-chloro- N -1-naphthylpropionamide, synthesized previously, was
added. Stirring was continued at 353 K for 2 h and then the mixture
was poured into cooled water. The brown precipitate was dried in an
oven at 313 K and recrystallized from ethanol as brown needles of (I).

Crystal data

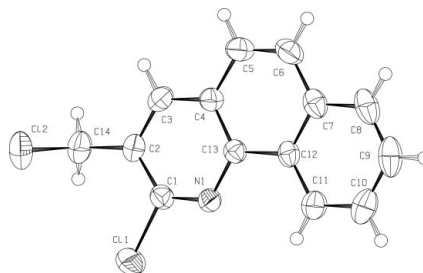
 $\text{C}_{14}\text{H}_9\text{Cl}_2\text{N}$
 $M_r = 262.12$
Triclinic, $P\bar{1}$
 $a = 6.9211$ (10) Å
 $b = 8.3176$ (10) Å
 $c = 10.6480$ (10) Å
 $\alpha = 79.21$ (10)°
 $\beta = 76.19$ (10)°
 $\gamma = 85.84$ (10)°
 $V = 584.5$ (3) Å³ $Z = 2$
 $D_x = 1.489$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 25
reflections
 $\theta = 2.5$ – 30.0 °
 $\mu = 0.53$ mm⁻¹
 $T = 293$ (2) K
Needle, brown
 $0.13 \times 0.05 \times 0.04$ mm

Figure 1

Molecular structure of (I). Displacement ellipsoids are shown at the 50%
probability level.

Data collection

Enraf–Nonius CAD-4
diffractometer
 $\theta/2\theta$ scans
Absorption correction: none
3498 measured reflections
3288 independent reflections
2022 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.085$

$\theta_{\text{max}} = 30.0^\circ$
 $h = -9 \rightarrow 9$
 $k = 0 \rightarrow 11$
 $l = -14 \rightarrow 14$
2 standard reflections
frequency: 120 min
intensity decay: 2%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.060$
 $wR(F^2) = 0.210$
 $S = 1.08$
3288 reflections
154 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0861P)^2 + 0.6126P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.41 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.35 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

C1–C1	1.756 (3)	N1–C1	1.302 (4)
C12–C14	1.799 (4)	N1–C13	1.354 (4)
C1–N1–C13	117.1 (3)	N1–C1–C2	126.6 (3)
N1–C13–C4	122.2 (3)	N1–C1–C11	115.0 (2)
N1–C13–C12	118.0 (3)	C2–C1–C11	118.4 (2)

All H atoms in (I) were positioned geometrically and allowed to ride on their parent atoms [$\text{C–H} = 0.93$ and 0.97 \AA , and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$].

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms, 1995); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON98* (Spek, 1998); software used to prepare material for publication: *SHELXL97*.

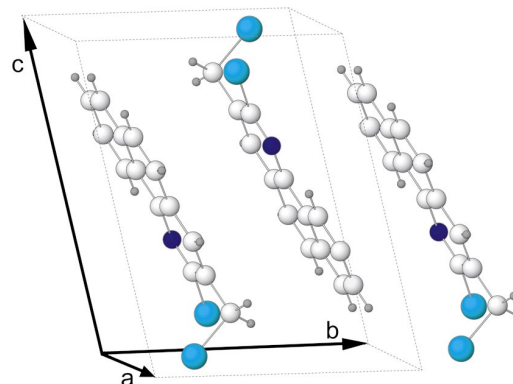


Figure 2
Packing diagram, showing the stacking of molecules of (I).

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